

8P1 Knowledge Organiser - Heating and Cooling

Key words:

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Thermal	A material that will let heat flow through it		
conductor			
Thermal	A material that will not let met flow		
insulator	through it.		
Conduction	Heat is conducted due to particles		
	vibrating and hitting each other. The		
	movement of heat (or electricity) through		
	a substance.		
Convection	The transfer of heat through a liquid or		
	gas (fluid) Convection occurs when		
	particles with a lot of heat energy in a		
	liquid or gas move and take the place of		
	particles with less heat energy		
Radiation	Heat can be transferred by infrared		
	radiation, this is an electromagnetic wave		
	and doesn't use particles.		
Temperature	temperature is a measure of how hot		
	something is.		
Heat	heat is a measure of the thermal energy		
	contained in an object.		
Thermal	Energy that is due to particles moving and		
energy	results in an object having a temperature.		
	It is transferred as heat		

Conduction: Conduction can only happen in **solids** as the particle must be toughing to pass on energy to it's neighbours.

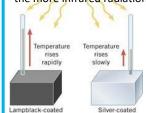
Heat moves from the hotter part of the object to the colder part

Particles in the metal are packed closely together. As they are heated the particles

gain kinetic energy and vibrate more. The particles that are vibrating collide with other particles and start to make them vibrate. This passes the kinetic energy from the heated particles to the cooler particles causing them to heat up too.

Radiation: Radiation doesn't need particles for the energy to Travel though as it is a type of electromagnetic radiation called infrared radiation.

Infrared radiation involves waves instead of particles. As such it can travel through a vacuum e.g. space. The hotter an object is, the more infrared radiation it emits.



block

You can experience radiation for yourself, on a warm day dull dark objects feel warmer as they **absorb** the thermal energy from the sun whereas shiny or white objects **reflect** the thermal energy and so feel cooler.

molecules in solid objects don't "move" - they vibrate or "jiggle"

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heat conducts from warm to cold

wire or other

t dip too.

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Convection

Convection

Water cools, becomes more dense, sinks

Particles with lots of heat energy in a liquid or gas move and take the place of particles with a lot of energy. Heat energy is

other particles they collide with.

Slower Warmer particles transfer their energy to their surroundings and begin to move more slowly.

Convection: This occurs naturally in fluids (liquid or gas) as

the particles are free to move and pass the thermal energy to

Cooler water becomes denser and sinks to the bottom of

move closer

together

taking up

less space.

Cooler particles take the space of the warmer ones that have

Cool
water sinks

This makes them move faster and take up more space. They are now less dense.

Thermal energy from the heat source is

	Conduction	Convection	Radiation
Particles	Υ	Υ	N
Solids	Υ	N	Υ
Liquids	N	Υ	Υ
Gases	N	Υ	Υ
Particles move far part	N	Υ	n/a
Particles vibrate on the spot	Υ	N	n/a
Particles rise and fall to transfer	N	Υ	n/a
energy			
Particles hit each other to	Υ	N	n/a
transfer energy			



